Screening for Antioxidant Phenolic Compounds from *Polygonum* capitatum

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Abstract [Objectives] To discover antioxidant natural products from the famous Hmong medicinal plant *Polygonum capitatum*. [Methods] The antioxidant activities of the isolated components were evaluated by ABTS and DPPH assays. [Results] A total of 27 free phenolics were isolated form *P. capitatum*. Then the *in vitro* antioxidant potential of these components was evaluated according to the DPPH and ABTS radical scavenging assays. Among them, five compounds (13, 14, 17, 23, and 25) showed most significant ABTS radical-scavenging activity (*IC*₅₀ values of 3.81 – 15.09 μg/mL). And 12 components (1, 2, 6, 7, 9, 12, 13, 14, 16, 17, 23, and 25) showed notable radical scavenging activity against DPPH (inhibition rates >88%). [Conclusions] Most of the above bioactive compounds were reported for the first time.

Key words *Polygonum capitatum*, Phenolic compounds, Antioxidant activity

1 Introduction

Polygonum capitatum (Touhualiao in Chinese), a well-known and large-scale Miao medicinal plant in China, was derived from *P. capitatum* (Buch. -Ham. ex D. Don) H. Gross and is widely used for the treatment of urinary tract infections, cystitis, eczema, pyelonephritis, lowering uric acid and anti-gout^[1-2]. *P. capitatum* demonstrated a variety of pharmacological activities, including anti-inflammatory, antioxidant, antimicrobial, anticancer, analgesic and other bioactive effects^[3-4].

The antioxidant activities were attributed to the extracts of P. capitatum. The antioxidant activities of P. capitatum have been extensively studied using different anti-oxidant models. These models were induced by 2-20-azinobis-3-ethylben-zthia zoline-6-sulphonate (ABTS), 1, 1-diphenyl-2-picrylhydraz-yl (DP-PH), hydrogen peroxide (H₂O₂). Base on the literatures, P. capitatum extract has displayed obvious antioxidant activity in vitro. 80% methanol extract of leaves and stems from P. capitatum demonstrated strong antioxidant activities against ABTS +/ OH^{-[5]}. The ethanol extract revealed stronger antioxidant activities than the agueous extracts of P. capitatum^[6]. The same result was shown in another study, the methanol extract of P. capitatum revealed higher scavenging activity against DPPH radical and ABTS radical^[7]. Additionally, the EtOAc extract of P. capitatum exhibited remarkable scavenging activity against DPPH and ABTS radicals. However, little is known about the chemical components

Received; March 2, 2024 Accepted; July 26, 2024
Supported by Guizhou Provincial Science and Technology [ZK (2022)-362, ZK (2024)-047], and (2023) ZK01]; the Innovation and Entrepreneurship Training Program for Undergraduates from China (202210660131 & 202310660082); Science Foundation of Guizhou Education Technology (2022-064); University Engineering Research Center for the Prevention and Treatment of Chronic Diseases by Authentic Medicinal Materials in Guizhou Province ([2023]035); Science and Technology Research Project of Guizhou Administration of Traditional Chinese Medicine (QZYY-2024-134).

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from P. $capitatum^{[8]}$.

In this study, we intended to discover novel bioactive compounds and expand the diversity and production of bioactive natural products by extracting compounds from the whole herb *P. capitatum.* 27 compounds were isolated from the ethyl acetate extract of *P. capitatum.* And the antioxidant activities of the isolated components were evaluated by ABTS and DPPH assays.

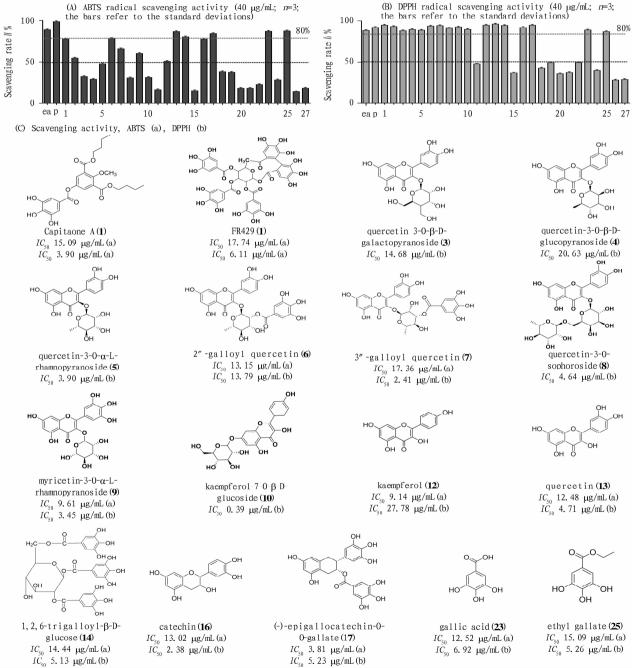
2 Results and analysis

2.1 Compounds structure elucidation The compounds (1-27) were capitaone A (1), FR429 (davidiin, 2), quercetin-3-O- β -D-galactopyranoside (3), quercetin-3-O- β -D-glucopyranoside (4), quercetin-3-O- α -L-rhamnopyranoside (5), 2"-galloylquercitrin (6), 3"-galloylquercitrin (7), quercetin-3-O-sophoroside (8), quercetin-3-O- α -L-rhamnopyranoside (9), kaempferol-7-O- β -D-glucopyranoside (10), kaempferol-3-O- α -L-rhamnoside (11), kaempferol (12), quercetin (13), 1, 2, 6-trigalloyl- β -D-glucose (14), stigmast-5-en-3-O- β -D-glucoside (15), catechin (16), (-)-epigallocatechin-3-O-gallate (17), silybin (18), apigenin (19), alpine isoflavones (20), 3, 3', 4'-trimethyl ellagic acid (21), 5, 7-dihydroxy-4H-4-chromone (22), gallic acid (23), vanillic acid (24), ethyl gallate (25), β -sitosterol (26) and methyl arachidonate (27) [9].

2.2 Antioxidant activity According to several studies, the main factor causing degenerative diseases was free radicals, which were actively produced through the human body metabolic process. The increase in the number of free radicals in the body can lead to the disruption of the body's metabolic processes, which can lead to various disease-causing processes. In this study, the antioxidant capacity of compounds 1 – 27 (40 $\mu g/mL$) was assessed using two antioxidant assays (ABTS, DPPH). As shown in Fig. 1, 12 compounds (1, 2, 6, 7, 9, 12 – 14, 16, 17, 23, 25) exhibited strong antioxidant activity with inhibition rates > 50% for the ABTS test. Among them, 9 (myricetin-3-0- α -L-rhamnopyranoside), 12 (kaempferol) and 17 ((–)-epigallocatechin-

3-O-gallate)) were the most active ones with an IC_{50} value from 3.81 to 9.61 µg/mL. Compounds 2 (FR429, IC_{50} 17.74 µg/mL), 7 (3"-galloyl quercitrin, IC_{50} value 17.38 µg/mL) and 14 (1, 2, 6-trigalloyl-β-D-glucose, IC_{50} value 14.44 µg/mL) were reported here for the first time. The activities of 6 (2"-galloyl quercitrin), 9 (myricetin-3-O-α-L-rhamnopyranoside), 12 (kaempferol), 13 (quercetin), 16 (catechin), 17 (–)-epigallocatechin-3-O-gallate), 23 (gallic acid), 25 (ethyl gallate) were consistent with previous reports.

Interestingly, all of the four active compounds $(\mathbf{5} - \mathbf{7}, \mathbf{9})$ possess one quercitrin group, and compounds $(\mathbf{6}, \mathbf{7}, \mathbf{9})$ possessed an extra gallic or hydroxyl group more than the compound 5. For instance, compound 5 showed a scavenging rate of only 48%, whereas its galloyl derivative 6 was 78%. Similar results were also observed for $\mathbf{5}$ $(48\%)/\mathbf{7}$ (66%), and $\mathbf{5}$ $(48\%)/\mathbf{9}$ (61%). And then although the only difference between compound $\mathbf{12}$ and compound $\mathbf{13}$ is an additional hydroxyl group, inhibition rate was better for compound $\mathbf{13}$ (87%) than compound $\mathbf{12}$ (51%), (Fig. 1).



NOTE (A) ABTS radical scavenging activity; (B) DPPH radical scavenging activity; (C) The dose-inhibition rate curves for potent compounds and their IC₅₀ values.

Fig. 1 Antioxidant capacity of compounds 1 - 27 from Polygonum capitatum

For the DPPH radical scavenging assay, the EtOAc extract exhibited stronger inhibitory activities (inhibition rate 88.6%, 40 μ g/mL). Among the them, 17 ones (1 – 10, 12 – 14, 16, 17, 23, 25) showed remarkable DPPH radical scavenging activity,

with inhibition rates ranging from 88.2% to 96.3% at 40 μ g/mL. Compounds **2**, **6**, and **7** were reported here for the first time, with an IC_{50} value of 6.11, 13.79 and 2.41 μ g/mL, respectively (Fig. 2).

Fig. 2 Structure-activity relationship analysis of the active compounds for DPPH experiment

2.3 ABTS radical scavenging assay The method ABTS has been commonly used to determine the antioxidant capacity of the biological samples. In brief, the ABTS free radical manucaption solution (ABTS⁺) was generated by reacting ABTS solution (7.0 mM) with potassium persulfate (2.45 mM) to produce bluegreen color solution 24 h^[10]. The more obvious the disappearance of blue-green colour solution, the stronger the antioxidant activity. To evaluate the antioxidant activity, the ABTS + solution was diluted with ethanol until its absorbance reached 0.70 \pm 0.02 at the wavelength of 734 nm. Therefore, the ABTS + solution was obtained. Then, the EtOAc extract and the components (100 µL, 40 µg/mL) with ABTS * solution (100 µL) were added to the 96well plates (as the value of A_i) and were interacted for 10 min at room temperature in the dark. Immediately followed by the absorbance at 734 nm was measured [11]. Vitamin C (40 µg/mL) as the positive control. The percentage of radical scavenging was calculated using the Formula (1). Herein, A_i represents the absorbance of the sample, A, denotes the value of 100 µL compounds with 100 μ L ethanol, and A_0 is the absorbance of the control solution.

Radical scavenging activity (%) = $[1 - (A_i - A_j)/A_0] \times 100\%$ (1)

2.4 DPPH radical scavenging assay Antioxidant properties of extracts obtained from P. capitatum were determined using the DP-PH method [12]. The antioxidant effect on interaction with DPPH, transfers electron or hydrogen atom to DPPH, thus neutralizing the free radical character and converting it to 1, 1-diphenyl-2-picryl-hydrazine. Briefly, 4.0 mg DPPH powder was dissolved with 100 mL ethanol to obtain the DPPH solution (0.01 mM) in a volumetric flask. Then, the EtOAc and the compounds (40 µg/mL) with DPPH solution (100 µL) (A_i) were added to 96-well plates and incubated for 30 min at room temperature. Subsequently, the activity was measured using the Thermo Scientific Varioskan LUX

(Berthold, VL0L00D0, USA), with an absorbance of 570 nm and the data was calculated according to the Formula (1). Also, vitamin C (40 μ g/mL) was used as the positive control.

3 Conclusions

In this study, tighter with 27 compounds were isolated and identified of the EtOAc extract from P. capitatum. 5 compounds (13, 14, 17, 23, 25) exhibited notable radical scavenging activity against ABTS (IC_{50} values 12.48, 14.44, 3.81, 12.52 and 15.09 μ g/mL). In particular, 12 components (1, 2, 6, 7, 9, 12 – 14, 16, 17, 23, and 25) showed remarkable DPPH radical scavenging activity, and the inhibition rates were in the range of 88.2% to 96.3%. They may be promising natural agents of the antioxidant.

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